POLYSACCHARIDES WITH REACTIVE CARBONYLS: CHEMICAL AND PHYSICAL PROPERTIES

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Polysaccharides may contain reactive keto- or aldehydo groups, mostly attributed to oxidative side reactions during processing, and in some cases arising from specific chemical or enzymatic reactions designed for the purpose of oxidation, for instance periodate. In addition to chemical reactivity, the introduction of carbonyl groups directly or indirectly influence the physical properties, for instance chain flexibility.^{1,2}

carbonyl profiles of The a pectin ('sphagnan')³ from Sphagnum moss and some periodate oxidized polysaccharides were assayed by fluorescent labeling or reduction with tritiated borohydride, in combination with multidetector SEC. The tritium incorporation method was preferred for alkali stable polysaccharides, while the CCOA method was most suitable for acid stable polysaccharides with low carbonyl content. The 2-AB method is applicable for all polysaccharides tested with varying carbonyl content; however it lacks the ability to detect ketone functionalities⁴. Interestingly, periodate oxidized alginate reacted only with one molecule of 2-AB per dialdehyde. Sphagnan displayed a special carbonyl profile, with the highest carbonyl content at high molecular weights.

Recently, periodate oxidation was combined with in vitro epimerization of mannuronan – a special alginate obtained from an epimerase-negative strain of *Pseudomonas*. In this way, novel alginates were obtained⁵. Some of their chemical and gelling properties will be discussed.

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